Chem 532: Problem Set #4

Due in class: Monday, Oct. 17

(1) A particle in a spherically symmetric potential is known to be in an eigenstate of ℓ^2 and ℓ_z with eigenvalues $\hbar^2 \ell (\ell + 1)$ and $m\hbar$, respectively. Show that the expectation values between

 $|\ell m\rangle$ eigenstates satisfy

$$\left\langle \ell_x \right\rangle = \left\langle \ell_y \right\rangle = 0$$
 and
 $\left\langle \ell_x^2 \right\rangle = \left\langle \ell_y^2 \right\rangle = \frac{\left[\ell(\ell+1)\hbar^2 - m^2\hbar^2\right]}{2}$

(2) Using the real-valued spherical harmonics defined as

$$Y_{x} = \frac{1}{\sqrt{2}} \left(Y_{1,-1} - Y_{1,1} \right)$$
$$Y_{y} = \frac{i}{\sqrt{2}} \left(Y_{1,1} + Y_{1,-1} \right)$$
$$Y_{z} = Y_{1,0}$$

evaluate each of the following:

- (a) $\langle Y_{y} | \ell_{y} | Y_{y} \rangle$ (b) $\langle Y_{y} | \ell_{z} | Y_{y} \rangle$ (c) $\langle Y_{z} | \ell_{x} | Y_{z} \rangle$ (d) $\langle Y_{z} | \ell_{x}^{2} + \ell_{y}^{2} | Y_{z} \rangle$
- (3) Locate the radial nodes of the 3*s* orbital of the hydrogen atom. How do these compare to those of Ar^{17+} ?

(4) Calculate (a) the mean radius, (b) the mean square radius, and (c) the most probable radius of the 2s orbital of a hydrogenic atom of atomic number Z. You will find the following integral useful:

$$\int_{0}^{\infty} x^{n} e^{-ax} dx = \frac{n!}{a^{n+1}}$$