

Table of Useful Integrals, etc.

$$\int_0^{\infty} e^{-ax^2} dx = \frac{1}{2} \left(\frac{\pi}{a} \right)^{1/2}$$

$$\int_0^{\infty} x e^{-ax^2} dx = \frac{1}{2a}$$

$$\int_0^{\infty} x^2 e^{-ax^2} dx = \frac{1}{4a} \left(\frac{\pi}{a} \right)^{1/2}$$

$$\int_0^{\infty} x^3 e^{-ax^2} dx = \frac{1}{2a^2}$$

$$\int_0^{\infty} x^{2n} e^{-ax^2} dx = \frac{1 \cdot 3 \cdot 5 \cdots (2n-1)}{2^{n+1} a^n} \left(\frac{\pi}{a} \right)^{1/2}$$

$$\int_0^{\infty} x^{2n+1} e^{-ax^2} dx = \frac{n!}{2a^{n+1}}$$

$$\int_0^{\infty} x^n e^{-ax} dx = \frac{n!}{a^{n+1}}$$

Integration by Parts:

$$\int_a^b U dV = [UV]_a^b - \int_a^b V dU \quad U \text{ and } V \text{ are functions of } x. \text{ Integrate from } x = a \text{ to } x = b$$

$$\int \sin(ax) dx = -\frac{1}{a} \cos(ax)$$

$$\int \sin^2(ax) dx = \frac{x}{2} - \frac{\sin(2ax)}{4a}$$

$$\int \sin^3(ax) dx = -\frac{1}{a} \cos(ax) + \frac{1}{3a} \cos^3(ax)$$

$$\int \sin^4(ax) dx = \frac{3x}{8} - \frac{3\sin(2ax)}{16a} - \frac{\sin^3(ax)\cos(ax)}{4a}$$

$$\int \sin(ax)\sin(bx) dx = \frac{\sin[(a-b)x]}{2(a-b)} - \frac{\sin[(a+b)x]}{2(a+b)} \quad \text{where } a^2 \neq b^2$$

$$\int \cos(ax)\cos(bx) dx = \frac{\sin[(a-b)x]}{2(a-b)} + \frac{\sin[(a+b)x]}{2(a+b)}$$

$$\int \sin(ax)\cos(bx)dx = \frac{-\cos[(a-b)x]}{2(a-b)} - \frac{\cos[(a+b)x]}{2(a+b)}$$

$$\int x \sin^2(ax)dx = \frac{x^2}{4} - \frac{x \sin(2ax)}{4a} - \frac{\cos(2ax)}{8a^2}$$

$$\int x^2 \sin^2(ax)dx = \frac{x^3}{6} - \left(\frac{x^2}{4a} - \frac{1}{8a^3}\right) \sin(2ax) - \frac{x \cos(2ax)}{4a^2}$$

$$\int x \sin(ax)\sin(bx)dx = \frac{\cos[(a-b)x]}{2(a-b)^2} - \frac{\cos[(a+b)x]}{2(a+b)^2} + \frac{x \sin[(a-b)x]}{2(a-b)^2} - \frac{x \sin[(a+b)x]}{2(a+b)^2}$$

$$\int x \sin(ax)dx = \frac{\sin(ax)}{a^2} - \frac{x \cos(ax)}{a}$$

$$\int x \cos(ax)dx = \frac{x \sin(ax)}{a} + \frac{\cos(ax)}{a^2}$$

$$\int \cos(ax)dx = \frac{\sin(ax)}{a}$$

$$\int \cos^2(ax)dx = \frac{x}{2} + \frac{\sin(2ax)}{4a}$$

$$\int x^2 \cos^2(ax)dx = \frac{x^3}{6} + \left(\frac{x^2}{4a} - \frac{1}{8a^3}\right) \sin(2ax) + \frac{x \cos(2ax)}{4a^2}$$

$$\int \cos(bx)e^{-ax^2} dx = \frac{e^{ax}}{(a^2 + b^2)} [a \cos(bx) + b \sin(bx)]$$

Taylor Series:

$$\sum_{n=0}^{\infty} \frac{f^{(n)}(x_0)}{n!} (x - x_0)^n$$

Geometric Series:

$$\sum_{n=0}^{\infty} x^n = \frac{1}{1-x}$$

Euler's Formula:

$$e^{i\phi} = \cos \phi + i \sin \phi$$

Quadratic Equation and other higher order polynomials:

$$ax^2 + bx + c = 0$$

$$x = \frac{-b \pm \sqrt{b^2 - 4ac}}{2a}$$

$$ax^4 + bx^2 + c = 0$$

$$x = \pm \left(\frac{-b \pm \sqrt{b^2 - 4ac}}{2a} \right)$$

General Solution for a Second Order Homogeneous Differential Equation with Constant Coefficients:

$$\text{If: } y'' + py' + qy = 0$$

Assume a solution for y:

$$y = e^{sx} \quad y' = se^{sx} \quad y'' = s^2 e^{sx}$$

$$\therefore s^2 e^{sx} + pse^{sx} + qe^{sx} = 0$$

$$\text{and } s^2 + ps + q = 0$$

$$\text{Hence } y = c_1 e^{s_1 x} + c_2 e^{s_2 x}$$

Conversions from spherical polar coordinates into Cartesian coordinates:

$$x = r \sin \theta \cos \phi$$

$$y = r \sin \theta \sin \phi$$

$$z = r \cos \theta$$

$$dv = r^2 \sin \theta dr d\theta d\phi$$

$$0 < r < \infty$$

$$0 < \theta < \pi$$

$$0 < \phi < 2\pi$$

Commutator Identities:

$$[\hat{A}, \hat{B}] = -[\hat{B}, \hat{A}]$$

$$[\hat{A}, \hat{A}^n] = 0 \quad n = 1, 2, 3, \dots$$

$$[k\hat{A}, \hat{B}] = [\hat{A}, k\hat{B}] = k[\hat{A}, \hat{B}]$$

$$[\hat{A} + \hat{B}, \hat{C}] = [\hat{A}, \hat{C}] + [\hat{B}, \hat{C}]$$

$$[\hat{A}, \hat{B}\hat{C}] = [\hat{A}, \hat{B}]\hat{C} + \hat{B}[\hat{A}, \hat{C}]$$

$$[\hat{A}\hat{B}, \hat{C}] = [\hat{A}, \hat{C}]\hat{B} + \hat{A}[\hat{B}, \hat{C}]$$

Creation and Annihilation Operators

$$l_{\pm} |l, m_l, s, m_s\rangle = (l(l+1) - m_l(m_l \pm 1))^{1/2} \hbar |l, m_{l\pm 1}, s, m_s\rangle$$

$$s_{\pm} |l, m_l, s, m_s\rangle = (s(s+1) - m_s(m_s \pm 1))^{1/2} \hbar |l, m_l, s, m_{s\pm 1}\rangle$$

$$j_{\pm} |j, m_j\rangle = (j(j+1) - m_j(m_j \pm 1))^{1/2} \hbar |j, m_{j\pm 1}\rangle$$

Atomic Units:

Quantity	Atomic unit in cgs or other units	Values of some atomic properties in atomic units (a.u.)
Mass	$m_e = 9.109534 \times 10^{-28} \text{ g}$	Mass of electron = 1 a.u.
Length	$a_0 = 4\pi\epsilon_0\hbar^2/m_e e^2$ $= 0.52917706 \times 10^{-10} \text{ m}$ (= 1 bohr)	Most probable distance of 1s electron from nucleus of H atom = 1 a.u.
Time	$\tau_0 = a_0\hbar/e^2$ $= 2.4189 \times 10^{-17} \text{ sec}$	Time for 1s electron in H atom to travel one bohr = 1 a.u.
Charge	$e = 4.803242 \times 10^{-10} \text{ esu}$ $= 1.6021892 \times 10^{-19} \text{ coulomb}$	Charge of electron = -1 a.u.
Energy	$e^2/4\pi\epsilon_0 a_0 = 4.359814 \times 10^{-18} \text{ J}$ (= 27.21161 eV \equiv 1 hartree)	Total energy of 1s electron in H atom = -1/2 a.u.
Angular momentum	$\hbar = h/2\pi$ $= 1.0545887 \times 10^{-34} \text{ J sec}$	Angular momentum for particle in ring = 0, 1, 2, ... a.u.
Electric field strength	$e/a_0^2 = 5.1423 \times 10^9 \text{ V/cm}$	Electric field strength at distance of 1 bohr from proton = 1 a.u.

Physical Constants:

Values of Some Physical Constants

Constant	Symbol	Value
Atomic mass constant	m_u	$1.660\,5402 \times 10^{-27}$ kg
Avogadro constant	N_A	$6.022\,1367 \times 10^{23}$ mol ⁻¹
Bohr magneton	$\mu_B = e\hbar/2m_e$	$9.274\,0154 \times 10^{-24}$ J · T ⁻¹
Bohr radius	$a_0 = 4\pi\epsilon_0\hbar^2/m_e e^2$	$5.291\,772\,49 \times 10^{-11}$ m
Boltzmann constant	k_B	$1.380\,658 \times 10^{-23}$ J · K ⁻¹ 0.695 038 cm ⁻¹
Electron rest mass	m_e	$9.109\,3897 \times 10^{-31}$ kg
Gravitational constant	G	$6.672\,59 \times 10^{-11}$ m ³ · kg ⁻¹ · s ⁻²
Molar gas constant	R	$8.314\,510$ J · K ⁻¹ · mol ⁻¹ 0.083 1451 dm ³ · bar · K ⁻¹ · mol ⁻¹ 0.082 0578 dm ³ · atm · K ⁻¹ · mol ⁻¹
Molar volume, ideal gas (one bar, 0°C)		22.711 08 L · mol ⁻¹
(one atm, 0°C)		22.414 09 L · mol ⁻¹
Nuclear magneton	$\mu_N = e\hbar/2m_p$	$5.050\,7866 \times 10^{-27}$ J · T ⁻¹
Permittivity of vacuum	ϵ_0 $4\pi\epsilon_0$	$8.854\,187\,816 \times 10^{-12}$ C ² · J ⁻¹ · m ⁻¹ $1.112\,650\,056 \times 10^{-10}$ C ² · J ⁻¹ · m ⁻¹
Planck constant	h \hbar	$6.626\,0755 \times 10^{-34}$ J · s $1.054\,572\,66 \times 10^{-34}$ J · s
Proton charge	e	$1.602\,177\,33 \times 10^{-19}$ C
Proton magnetogyric ratio	γ_p	$2.675\,221\,28 \times 10^8$ s ⁻¹ · T ⁻¹
Proton rest mass	m_p	$1.672\,6231 \times 10^{-27}$ kg
Rydberg constant (Bohr)	$R_\infty = m_e e^4/8\epsilon_0^2 h^2$	$2.179\,8736 \times 10^{-18}$ J 109 737.31534 cm ⁻¹
Rydberg constant (exptl)	R_H	109 677.581 cm ⁻¹
Speed of light in vacuum	c	299 792 458 m · s ⁻¹ (defined)
Stefan–Boltzmann constant	$\sigma = 2\pi^5 k_B^4/15h^3 c^2$	$5.670\,51 \times 10^{-8}$ J · m ⁻² · K ⁻⁴ · s ⁻¹

Operators:

TABLE 4.1

Classical-mechanical observables and their corresponding quantum-mechanical operators.

Name	Observable		Operator
	Symbol	Symbol	Operation
Position	x	\hat{X}	Multiply by x
	\mathbf{r}	$\hat{\mathbf{R}}$	Multiply by \mathbf{r}
Momentum	p_x	\hat{p}_x	$-i\hbar \frac{\partial}{\partial x}$
	\mathbf{p}	$\hat{\mathbf{P}}$	$-i\hbar \left(\mathbf{i} \frac{\partial}{\partial x} + \mathbf{j} \frac{\partial}{\partial y} + \mathbf{k} \frac{\partial}{\partial z} \right)$
Kinetic energy	K_x	\hat{K}_x	$-\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2}$
	K	\hat{K}	$-\frac{\hbar^2}{2m} \left(\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} \right)$ $= -\frac{\hbar^2}{2m} \nabla^2$
Potential energy	$V(x)$	$\hat{V}(\hat{x})$	Multiply by $V(x)$
	$V(x, y, z)$	$\hat{V}(\hat{x}, \hat{y}, \hat{z})$	Multiply by $V(x, y, z)$
Total energy	E	\hat{H}	$-\frac{\hbar^2}{2m} \left(\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} \right)$ $+ V(x, y, z)$ $= -\frac{\hbar^2}{2m} \nabla^2 + V(x, y, z)$
Angular momentum	$L_x = yp_z - zp_y$	\hat{L}_x	$-i\hbar \left(y \frac{\partial}{\partial z} - z \frac{\partial}{\partial y} \right)$
	$L_y = zp_x - xp_z$	\hat{L}_y	$-i\hbar \left(z \frac{\partial}{\partial x} - x \frac{\partial}{\partial z} \right)$
	$L_z = xp_y - yp_x$	\hat{L}_z	$-i\hbar \left(x \frac{\partial}{\partial y} - y \frac{\partial}{\partial x} \right)$